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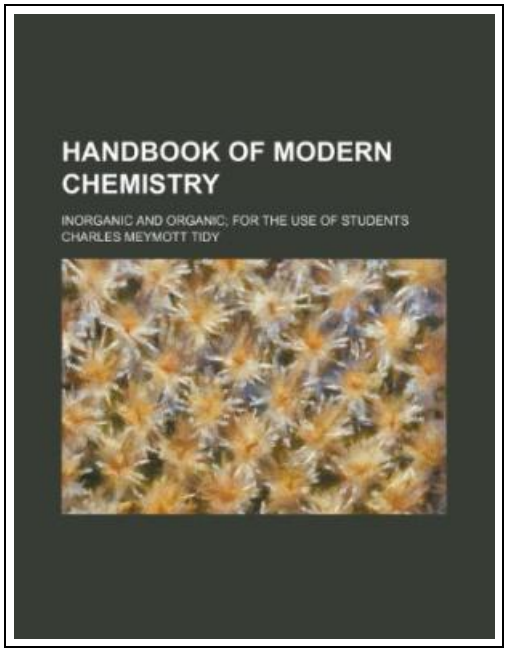

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Reviews

Good e book and helpful one. It is really basic but excitement from the 50 % of your pdf. Your way of life span is going to be enhance when you comprehensive looking at this pdf.
(Novella Maggio)

HANDBOOK OF MODERN CHEMISTRY; INORGANIC AND ORGANIC FOR THE USE OF STUDENTS

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1878 Excerpt: . exposing a solution of the salt to air, a brown liquid results, due to the formation of a normal and basic ferric sulphate, the former remaining in solution and the latter being precipitated (10FeS04+Os =3(Fes03,3S03) + 2Fe203, SO,). A similar result occurs by the exposure of the solid crystals to air. The aqueous solution of the salt rapidly absorbs nitric oxide. It forms double salts (isomorphous with the magnesic salts), with potassic and amnionic sulphate. Uses.--In the laboratory it is used as a reducing agent, as, e. g., to precipitate metallic gold from its solutions. In the arts it is used in the manufacture of black dyes and of ink, by the action of tannic acid infusions upon it. (14-) Ferric Sulphate.--Persulphate or sesquisulphate of iron (Fe 23S04), is found native in Chili, as coquimbite (Fej304,9H20). Preparation.--By adding the necessary quantity of H2S04 to Fe S04 to convert it into Fe23S04. The mixture is then boiled, the iron being afterwards peroxided by the cautious addition of nitric acid. Properties.--A non-crystalline, yellow body. (15.) Ferrous Nitrate (Fe2N03,6H20) is prepared by the action of cold dilute nitric acid on ferrous sulphide; whilst (16.) Ferric Nitrate (Fe s6N03,12H20) is prepared by the action of nitric acid (specific gravity 12) on metallic iron. N.B.--Fuming Nitric Acid (specific gravity 1-5) has no action on iron. (See Passive Iron, p. 352.) (17.) FeiT0U8 Carbonate, Protocarbonate of iron (FeC03), is found native as spathic iron ore, and as clay iron ore....

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